

Single Particle Deformation and Analysis of Silica-Coated Gold Nanorods before and after Femtosecond Laser Pulse Excitation

Wiebke Albrecht,^{*,†} Tian-Song Deng,[†] Bart Goris,[‡] Marijn A. van Huis,[†] Sara Bals,[‡] and Alfons van Blaaderen^{*,†}

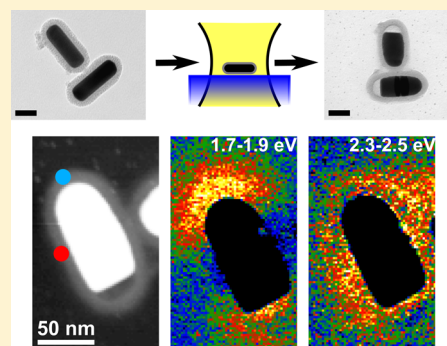
[†]Soft Condensed Matter, Debye Institute for Nanomaterials Science, Utrecht University, Princetonplein 5, 3584 CC Utrecht, The Netherlands

[‡]Electron Microscopy for Materials Research (EMAT), University of Antwerp, Groenenborgerlaan 171, 2020 Antwerp, Belgium

S Supporting Information

ABSTRACT: We performed single particle deformation experiments on silica-coated gold nanorods under femtosecond (fs) illumination. Changes in the particle shape were analyzed by electron microscopy and associated changes in the plasmon resonance by electron energy loss spectroscopy. Silica-coated rods were found to be more stable compared to uncoated rods but could still be deformed via an intermediate bullet-like shape for silica shell thicknesses of 14 nm. Changes in the size ratio of the rods after fs-illumination resulted in blue-shifting of the longitudinal plasmon resonances. Two-dimensional spatial mapping of the plasmon resonances revealed that the flat side of the bullet-like particles showed a less pronounced longitudinal plasmonic electric field enhancement. These findings were confirmed by finite-difference time-domain (FDTD) simulations. Furthermore, at higher laser fluences size reduction of the particles was found as well as for particles that were not completely deformed yet.

KEYWORDS: Gold nanorods, femtosecond laser excitation, deformation, EELS



Gold nanorods (NRs) have attracted much scientific attention due to their unique optical and photothermal properties.^{1–5} These result in a wide range of applications in various fields, such as photoacoustic imaging,^{6,7} catalysis,⁸ drug delivery,⁹ (bio)sensing,^{10–13} optical recording and data storage,^{14–17} and hyperthermic cancer treatment therapy in medicine.^{18,19} Several of these applications can even be combined using the same particles, like imaging and photothermal therapy²⁰ or image-guided drug delivery and hyperthermia,²¹ which makes gold nanorods even more versatile.

Localized surface plasmon resonances (LSPR) that are accompanied by efficient interactions of light with these nanoparticles that include strong local field enhancement of the particle tips lie at the core of many of those applications. Gold nanorods exhibit two distinct plasmon resonances, a transverse and a longitudinal one, because of their anisotropic shape and reduced symmetry compared to gold nanospheres. The transverse resonance, having a wavelength of approximately 520 nm, is the same as for spherical gold nanoparticles, whereas the longitudinal surface plasmon resonance depends strongly on the length to diameter aspect ratio of the nanorods. Therefore, it can be tuned to span the visible and near-infrared parts of the spectrum by changing the aspect ratio of the nanorods.^{22–24}

When exciting the longitudinal plasmon resonance with a high energy and short nanosecond (ns) or femtosecond (fs) laser pulse, the photon energy is absorbed by the electrons with subsequent electron–phonon scattering resulting in the heating

of the particle. This electron–phonon scattering takes on the order of 1–4 picoseconds (ps) for gold nanoparticles.²⁵ The heat is then released to the surrounding medium by diffusion, a process for which the time-scale strongly depends on the properties and geometry of the surrounding medium.²⁶ It is well established that gold nanorods undergo an aspect ratio reduction up to a complete deformation into a crystalline nanosphere or even fragmentation after absorption of sufficient photons.^{27–29} Furthermore, gold NRs can be deformed by thermal heating.^{30,31} The rate of heating and cooling and the final temperature of the particles are of crucial importance to the deformation mechanism and path followed. Petrova and co-workers inferred from their measurements that NRs that were deformed by fs-laser excitation did not show structural changes below 700 °C although the same particles already completely deformed at 250 °C when heated in an oven.³² These findings were attributed to the rods not staying hot for long enough during pulsed laser irradiation, indicating, as mentioned, that thermal diffusion plays an important role.

Most work so far aimed at understanding the light- and heat-induced deformation of Au nanorods was performed on large ensembles of these nanoparticles. Link et al. studied the dependence of the deformation on the laser pulse width and

Received: November 27, 2015

Revised: February 2, 2016

Published: February 12, 2016

found that significantly less pulse energy is needed to deform and melt the particles when using fs-pulses compared to ns-pulses.^{28,29} This was explained by competitive cooling during the pulse duration of a ns-pulse. At high laser fluences, fragmentation was observed. In related work, the same group measured a threshold energy of about 60 fJ to melt single nanorods with an average length and width of 44 and 11 nm, respectively, by fs-laser pulses in aqueous solution.³³

Zijlstra et al. showed that much information can be gained by performing single-particle measurements and thereby removing much of the heterogeneities of the sample.³⁴ They used white light scattering spectroscopy and scanning electron microscopy to study the laser-induced melting of single gold nanorods. They measured a melting energy of 260 fJ for nanorods with an average size of 92×30 nm and inferred from their measurements that nanorods with larger aspect ratio are thermodynamically less stable as they were found to deform more easily. Taylor et al. studied the reshaping behavior of individual nanorods under fs-laser illumination by scanning electron microscopy.³⁵ They concluded that the reshaping process is driven by curvature-induced surface diffusion of Au atoms over the nanorod surface and that the stability of the rods decreases with increasing aspect ratio.

From these results, it has become clear that correlating single-particle reshaping experiments with high-resolution electron microscopy and spectroscopy can reveal new insights into the atomistic mechanisms behind light-induced particle deformations. In our paper, we show additionally that if both single particle spectroscopy and single particle imaging can be combined on the same particle before and after deformation local shape and size effects can be connected to specific local plasmonic properties instead of to properties that are averaged over all particles. Examples of such properties are the flattening of the Au nanorods on only one side and a reduction in particle volume for some particles. Electron energy-loss spectroscopy (EELS) has recently proven to be a good technique for imaging local plasmonic properties of metallic nanoparticles. It has become possible to perform a local mapping of the plasmon resonances of single metal nanoparticles of various different shapes at a high resolution in energy and space,^{36–42} additionally coupling of resonances of particles that are near each other has been analyzed as well.^{36,37}

It is well-known that coatings surrounding gold nanorods can greatly influence the deformation behavior. Such coatings are often used to increase the colloidal stability of Au nanoparticles in solution as these metals are characterized by large Hamaker constants and thus large van der Waals attractions inducing aggregation. So far, mainly heat-induced particle deformation experiments were conducted to understand the influence of coatings on the deformation of gold nanorods.^{43,44} It was shown for instance that a thin carbon layer can drastically increase the thermal stability of a Au nanorod.⁴⁵ Silica is an important coating because it can enhance the thermal, mechanical, and colloidal stability. Furthermore, it can be made mesoporous and thus the particle's surface is still available for catalytic reactions and/or for sensing applications^{8,46–48} and, additionally, drugs can be efficiently transported due to a large surface area in the mesoporous shell.²¹

Surprisingly, however, little effort has been made to study the heat-induced deformation behavior of silica-coated gold nanorods. Recently, it was shown that the thermal stability for oven-heated rods can be drastically enhanced by a mesoporous silica coating.⁴⁹ Under illumination of ns-pulses,

mesoporous silica-coated gold nanorods were shown to be more stable as well.^{50,51} Also, for photoacoustic imaging silica-coated gold nanorods are promising candidates because coating the Au nanorods with silica improved the imaging properties.^{52,53} Finally, for cancer treatment therapies silica-coating of gold nanorods enhanced the performance. Zhang et al. showed that chemotherapeutic drug release was significantly improved and that hyperthermia is also possible for coated nanorods.²¹

Looking at the many possible application improvements of silica-coated gold nanorods, fundamental research on how such layers influence the deformation properties is needed. Single-particle measurements can help to unravel on a microscopic level the deformation process and the influence of a silica shell. As far as we know, these are the first single-particle deformation experiments of silica-coated gold nanorods under illumination of fs-laser pulses. The experiments were analyzed by transmission electron microscopy (TEM) and EELS. Our method allows for specifically deforming single particles in 3D assemblies,⁵⁴ which can lead to new applications like the modification of metallo-dielectric photonic crystals.

Results and Discussion. It is well-known that the position of the plasmon peak depends on the surrounding medium.^{1,25,55} Thus, in order to optimally excite the gold nanorods at their longitudinal plasmon resonance in our confocal setup the refractive index change from ethanol to glycerol needs to be taken into account. In glycerol, the longitudinal plasmon peak is red-shifted by about 30 nm to 860 nm (Figure S1) with respect to ethanol. For the deformation experiments, the excitation was therefore fixed at 860 nm but the power was varied. The laser power is directly connected to the temperature that the particle is heated to^{29,32,56} and thus the degree of deformation strongly depends on the applied laser power.

For different laser powers, the aspect ratio and shape of the same particles before and after laser irradiation were determined by TEM. By averaging over several particles for each applied laser power a general dependence on the degree of deformation was determined. Such an average is helpful to compare our measurements with previous studies. A typical outcome for different laser powers is shown in Figure 1b where the average aspect ratio of three different areas on the same TEM grid of silica-coated gold nanorods illuminated at three different values of the laser power was averaged over about 100 particles. At the lowest fluence of 5.3 mJ/cm^2 , the average aspect ratio decreased to about 3.0. With increasing fluence, the average aspect ratio was further reduced to about 2.6 for 8.1 mJ/cm^2 and to 1.6 for 10.6 mJ/cm^2 . The laser power that we used is generally much lower than the values that Chen et al. reported when using ns-laser pulses to deform silica-coated gold nanorods where changes in the absorption spectrum only occurred at 20 mJ/cm^2 for $34 \text{ nm} \times 9 \text{ nm}$ sized rods with a 20 nm thick silica shell.⁵⁰ This relates to the fact that fs-heating is more efficient than ns-heating as the total energy from the laser pulse can be transferred to the crystal before lattice cooling takes place. For uncoated gold nanorods, Link et al. also reported that fs-heating requires less laser power compared to ns-heating.^{28,29} In our experiment, many more laser pulses were supplied than needed for deformation as gold nanorods with an average size of $92 \text{ nm} \times 30 \text{ nm}$ but without a rigid shell could be even deformed by single fs-laser pulses with the degree of deformation being dependent on the fluence.³⁴ As we see a dependence on the provided fluence, the deformation process must be self-limiting in the sense that the laser light is not

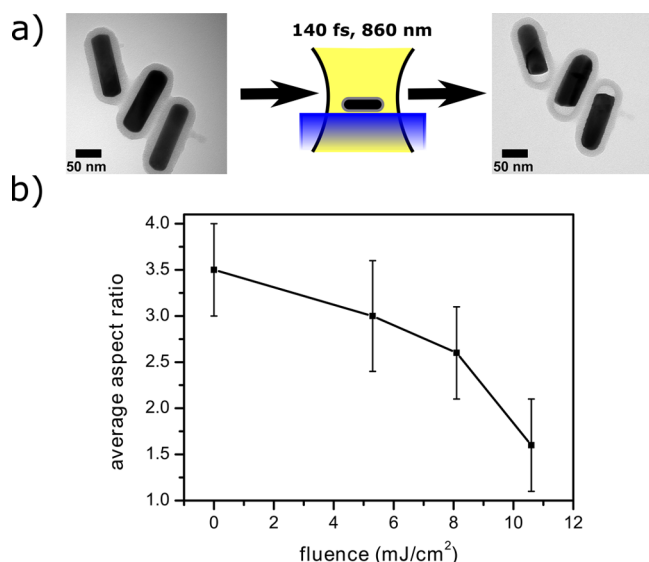


Figure 1. (a) The schematic experimental setup and typical TEM images of Au NRs before and after deformation. (b) The average aspect ratio after an applied laser power. The change in aspect ratio was averaged over about 100 particles. The initial average aspect ratio of the rods were 3.5 ± 0.5 coated with a 14 nm thick silica shell.

efficiently absorbed anymore once the aspect ratio drops below a certain value shifting the plasmon resonance out of the wavelength range of the laser excitation pulse. For example, at 5.3 mJ/cm^2 the average aspect ratio dropped to 3.0 leading to a longitudinal plasmon shift of about 25 nm (details Figure S1), which will thus be shifted out of the excitation pulse.

The curve in Figure 1b follows a similar trend as measurements from Zijlstra et al.³⁴ They performed single particle deformation experiments on gold nanorods without silica coating but embedded in a PVA film. For particles with an aspect ratio of 3 (92 nm × 30 nm), they found partial melting to start around 1.3 mJ/cm^2 and complete melting at 1.85 mJ/cm^2 . From that they calculated a melting energy of $Q_{\text{melt}} = 260 \text{ fJ}$ that is 15% higher than following thermodynamic considerations where they calculated Q_{melt} to be 225 fJ according to

$$Q_{\text{melt}} = \rho V [c_p (T_{\text{melt}} - T_0) + \Delta H_{\text{fus}}] \quad (1)$$

where ρ is the density of gold, V is the Au volume of the nanorod, c_p is the specific heat capacity of gold, T_{melt} is the melting temperature of gold, T_0 is the ambient temperature, and ΔH_{fus} is the heat of fusion for gold. The authors attributed the higher stability in experiments to the viscoelasticity of the PVA polymer matrix surrounding the particles, which hinders surface-driven migration, and to heat dissipation.

The energy Q_{abs} that is absorbed by a particle with an absorption cross section σ_{abs} under a laser fluence F is calculated by using

$$Q_{\text{abs}} = \sigma_{\text{abs}} F \quad (2)$$

The absorption cross section for a gold rod is available as an analytical expression from Mie-Gans theory for a spheroidal shape³²

$$\sigma_{\text{abs}} = \frac{2\pi V}{3\lambda} \epsilon_m^{3/2} \sum_j \frac{\left(\frac{1}{P_j}\right) \epsilon_2}{\left(\frac{\epsilon_1 + (1 - P_j) \epsilon_m}{P_j}\right)^2 + \epsilon_2^2} \quad (3)$$

where ϵ_1 and ϵ_2 are the real and imaginary part of the dielectric constant of gold, respectively, and ϵ_m is the dielectric constant of the medium. The depolarization factors P_j are given by

$$P_A = \left(\frac{1 - e^2}{e^2}\right) \left\{ \frac{1}{2e} \ln\left(\frac{1 + e}{1 - e}\right) - 1 \right\} \quad (4)$$

$$P_B = P_C = \frac{1 - P_A}{2} \quad (5)$$

where $e = \sqrt{1 - 1/(L/W)^2}$ is the eccentricity and L/W is the aspect ratio of the rod.

However, calculating a reliable absorption cross section is difficult because the analytical Mie-Gans theory, which is only available for particles with an ellipsoidal shape, does not accurately model experimental nanorods as not only the aspect ratio alone influences the extinction spectra but also end-cap geometries and the volume of the rods.⁵⁷ Prescott et al. showed that by adjusting the depolarization factors, reasonable results for the plasmon excitation wavelengths can be obtained (Figure S1). However, for obtaining absorption intensities experimental data are so far quantitatively more reliable. Zijlstra et al. measured experimental scattering cross sections for single gold nanorods with an average size of $5.8 \times 10^4 \text{ nm}^3$ and calculated the according absorption cross section by multiplying by the intensity ratio.³⁴ Using their values and using a correction factor of 1.7 to account for the volume difference, we can estimate Q_{abs} by $Q_{\text{abs}} = F(11.3 \times 10^{-11}) \text{ cm}^{-2}$. However, this represents the absorbed energy when the particle is oriented along the polarization of the laser beam. If the particle is oriented perpendicularly, it will absorb almost no energy. Our particles are randomly distributed with respect to the polarization and will thus only absorb about 1/2 of that energy. For the experimentally applied fluences of 5.3 mJ/cm^2 , 8.1 mJ/cm^2 , and 10.6 mJ/cm^2 the particles thus absorb approximately 300, 460, and 600 fJ on average. The required melting energy calculated by eq 1 is 380 fJ for our rods. Thus, for the highest applied laser fluence (where particles are still not all completely deformed to a spherical shape) the particles absorb about 55% more energy than required for melting, which is a lot more than the 15% that Zijlstra et al. measured for uncoated rods in PVA.

The increased stability is probably due to two reasons. First, heat diffusion is faster for silica-coated gold nanoparticles compared to particles without a silica shell. Hu et al. measured a decrease in the characteristic heat dissipation time constant after coating spherical gold nanoparticles with a thin silica shell in water compared to uncoated particles in water.²⁶ The characteristic heat dissipation time depends on the size of the particle and the thermal properties of the surroundings (details in SI). Hu et al. measured a time constant of 55 ps for a spherical gold particle with a thin silica shell in water. Taking different sizes and thermal properties of the solvent into account we roughly estimate a characteristic heat dissipation time of about 500 ps for our system (SI). The same estimation leads to a time constant of about 2000 ps for the particles in ref 34. Thus, heat gets dissipated faster away for our system. However, it was shown that structural transformations occur on fast time scales as well. Plech et al. inferred from their

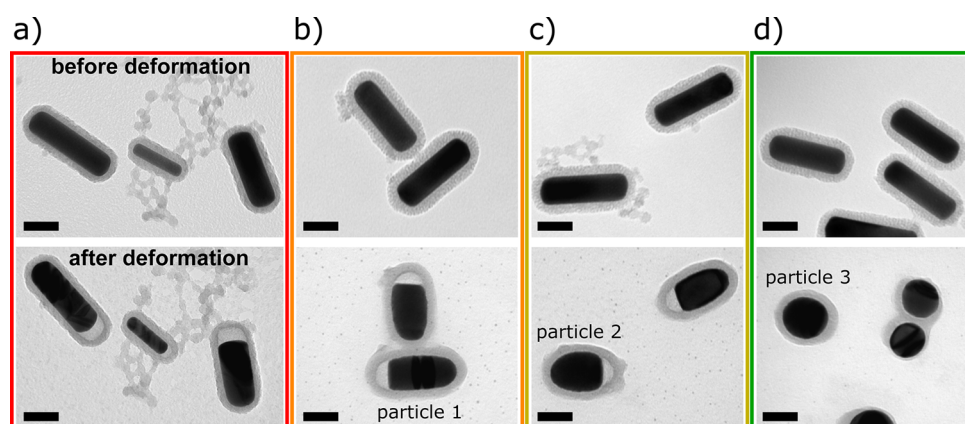


Figure 2. Comparison of the same particles before (upper row) and after (bottom row) deformation for average fluences of (a) 5.3 mJ/cm² and (b–d) 10.6 mJ/cm² (with differently absorbed energies, see text for clarification). The different columns show different stages of deformation. The absorbed energy increases hereby from left to right. The aspect ratios before and after deformation are summarized in Table 1. EELS measurements presented in Figure 3 were performed on the exact same particles marked with the name particle 1 to 3. All scale bars are 50 nm.

measurements that complete melting of 100 nm sized spherical particles in water (which is comparable in volume to our rods) occurred within 100 ps (their time-resolution limit).⁵⁸ Link et al. concluded from pump–probe absorption measurements that melting of micelle-capped gold nanorods in solution takes 30–35 ps.⁵⁹ Their nanorods were about 25 to 50 times smaller than ours. Thus, from literature it can be estimated that structural transformations for sizes of our rods can happen between 35 and 100 ps. Thus, heat dissipation is probably not the main reason for the increased stability.

Another influence of the silica shell is its rigidity, which hinders the migration of surface atoms and in addition hinders a shape change that does not fit in the original shell diameter. External oven-heating measurements performed by Gergely-Fülöp and co-workers confirmed that⁴⁹ as well as measurements performed in our group.⁶⁰ In their work, heat diffusion does not play a role due to the long time scales involved. For an aspect ratio of 3.4 (30 nm × 9 nm, 15 nm mesoporous silica shell) particles deformed to an almost spherical shape after heating at 700 °C for 1 h. In comparison, uncoated nanorods with an aspect ratio of 3.3 (73 nm × 22 nm) already completely deformed to spherical particles after heating at 250 °C for 1 h.³² Thus, the silica-coated nanorods were stable at much higher temperatures despite the significantly smaller volume and thus smaller surface to volume ratio. Therefore, it is likely that the enhanced stability for our rods is mostly influenced by the hindrance of surface atom movement and the confinement effects of the silica shell that has to deform to accommodate thicker rods.

The different absorbed energies are not only reflected in the changed aspect ratio but also in the shape of the resulting particles both with respect to the geometry of the gold core and silica shell. Figure 2 shows the same particles before (upper row) and after (bottom row) deformation where particles in different columns absorbed different energies (increasing from left to right). The corresponding aspect ratio changes are summarized in Table 1.

The amount of energy each single particle absorbs can vary strongly because it depends on the exact position of the particle with respect to the laser beam. Because the laser beam is scanned over the sample, some particles will end up right in the middle of the beam and thus absorb more energy as particles that are away from the middle. The *z*-position of the particle

Table 1. Aspect Ratios before and after Deformation for the Particles from Figure 2 Going from Left to Right

	a	b	c	d
before deformation	3.6	3.0	3.2	2.9
	4.3	3.0	3.5	2.9
	3.0			3.2
	3.0	1.6	1.3	1.0
after deformation	3.6	2.0	1.6	1.0
	2.2			1.0

also plays an important role here. The substrates are not perfectly flat and different particles will experience small differences in *z*-position. Because the laser beam goes through an objective with a high numerical aperture, different *z*-positions experience different powers. The intensity variation is expressed in the point spread function of the microscope used and has recently been measured for this system.⁶¹ For example, for 17 pJ pulse energy the fluence is 5.3 mJ/cm² in the focal plane and drops to 4.0 mJ/cm² for a *z*-position 240 nm below or above the focal plane (details in SI). Changing *z* might not have a significant influence because the particles already absorbed energy and consequently their longitudinal plasmon resonance shifted out of the excitation wavelength of the laser. Still, the sample was scanned in a small 3D stack (starting from the 2D focal plane determined in reflection mode). Another important factor that influences the amount of absorbed energy is the orientation of the NR with respect to the laser polarization. Particles that are aligned with the polarization will absorb more energy than particles that are not aligned.³⁵ Additionally, the polydispersity of the sample will influence the absorption behavior. If the longitudinal peak position of one particle does not optimally match the excitation wavelength it will absorb less energy. The above-mentioned points are reflected in the relatively large error bars, obtained from the standard deviation, observed in Figure 1. Hence, calculating a temperature that can be ascribed to a single particle is difficult and will be inaccurate, also because heat accumulation in the dispersion medium can play a role for high-repetition laser systems. Zijlstra et al. showed that the threshold to deform and damage particles increases with higher numerical aperture objectives.⁵² This was explained by the larger surface to volume ratio resulting in better heat dissipation for smaller focal spots.

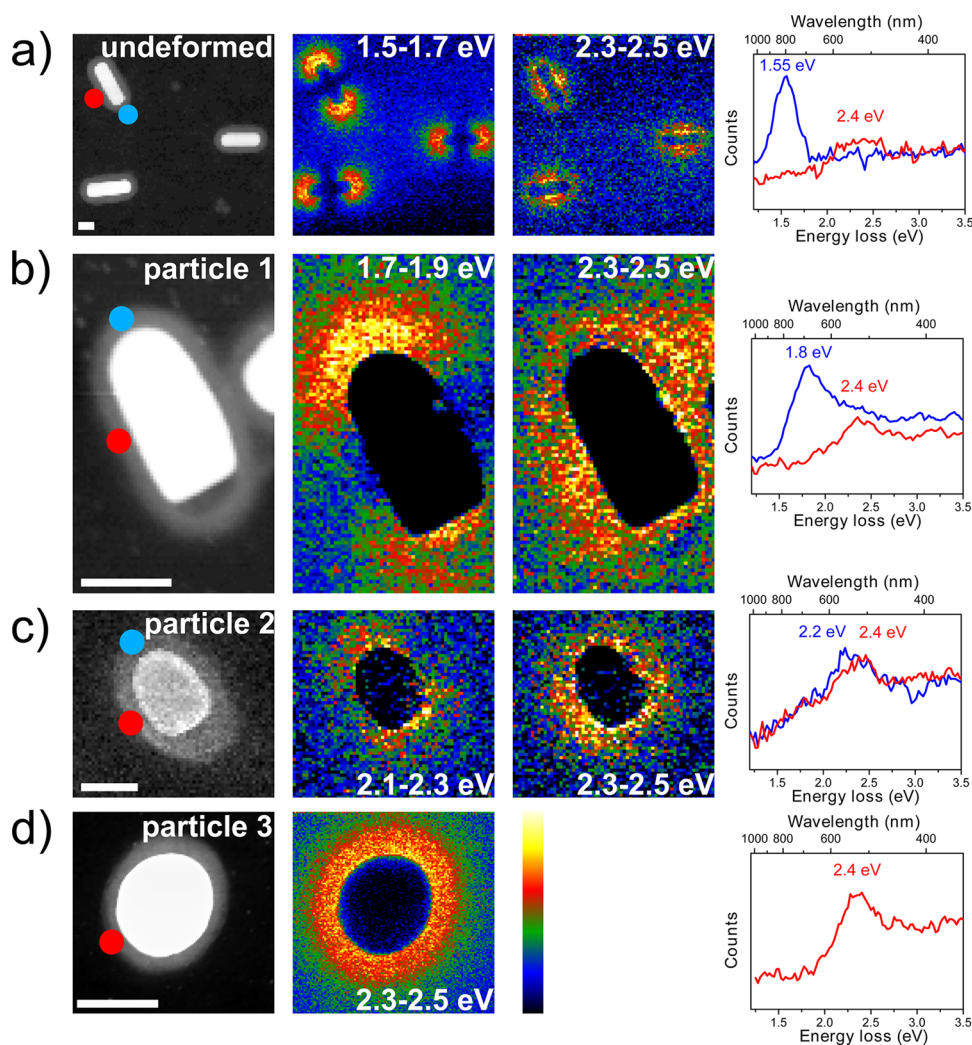


Figure 3. EELS measurements of undeformed (a), partially deformed (b,c) and completely deformed nanorods (d). The same particles that are shown in Figure 2 are presented here. The longitudinal plasmon resonance is blue-shifted with higher degree of deformation. The plasmon maps show the longitudinal and transverse character. All scale bars are 50 nm.

In our setup, heat accumulation might not play a major role because we have a high numerical aperture and thus a large surface to volume ratio of the focal spot.

Because of the considerations above, it cannot be determined which precise energy each single particle in Figure 2 absorbed but the amount of absorbed energy clearly increases from the left column to the right column because the deformation is enhanced. It can be seen that the particles deform from a rod shape via an intermediate bullet shape to a final spherical shape. Particles in column a only deformed slightly and adopt a rod shape that is shorter and wider than the original particles. Because the shell is mesoporous, it can most likely densify without changing its overall shape too much. The shape in the first column is mainly observed for the lowest used laser fluence of 5.3 mJ/cm^2 .

Particles in columns b–d were irradiated with an average fluence of 10.6 mJ/cm^2 but with different absorbed energies following the discussion above. The larger deformations could only be achieved by deforming the silica as well to make enough room for the larger width of the gold particle. A reduced mesoporosity can be observed as the mesopores are less visible in the TEM images after laser irradiation than before. Additionally, the shells slightly shrink. Because the shells

are mesoporous, shape changes due to the stress exerted by the deforming gold can occur more easily in comparison to a solid shell without internal pores. Furthermore, it was shown for ordered mesoporous silica that the pores shrink and the mesoporosity is reduced when heated above 600°C , a temperature that is likely reached in our experiments. The observed intermediate gold shapes, as in column b and c, were bullet-shaped. There is still room between the gold rod and the silica shell at the flat end of the particle. It is remarkable that in no case we observed a change in rod shape on both ends of the Au particles that remained flush with the silica shell on one end. An intermediate shape has not been so clearly and abundantly observed so far and as mentioned it is interesting that only one end was found to flatten and detach from the silica shell. Some groups reported intermediate ϕ -shapes although this was mainly observed for ns-pulses and particles without a solid shell.^{27,29} Chon et al. reported an intermediate dumbbell-like structure⁶² after fs-laser pulse illumination for initial rods with an aspect ratio of 4, a silica-shell thickness of 35 nm, and a smaller volume than our rods (no exact volume was provided but from the TEM image it can be estimated to be 15 times smaller than our volume). They argued that the structure occurred because the melting started at both tips of the rods

and stopped before a transformation to a spherical shape was completed. The exact intermediate shape most likely depends on the surface-to-volume ratio, shell thickness, or initial aspect ratio of the rod as well. Research to get more insight into what is important for what shape change is ongoing and the focus of the next paper. In the case of complete deformation to a spherical gold particle (column d), the silica is also completely deformed and was found to remarkably still enclose the spherical gold particles.

The changes in aspect ratio were accompanied by a change of the longitudinal plasmon resonance which we measured by EELS. First, we measured particles before laser irradiation which is shown in Figure 3a. The left image in Figure 3a is the dark-field image of the particles including two colored markers for the spots where the spectra were extracted that are shown in the right Figure of 3a. The spectrum recorded at the blue position shows a strong resonance at 1.55 eV (800 nm), which is the longitudinal plasmon resonance. The spectrum that was recorded at the red spot shows a peak at 2.4 eV (520 nm), which corresponds to the excitation of the transverse plasmon resonance. Comparing these measurements to the measured extinction spectrum in solution (Figure S1) one can see that the plasmon peak positions measured by EELS are blue-shifted. These shifts can be explained by the different dielectric media as the EELS measurements were performed in vacuum. When inserting the dielectric constant of vacuum into eq 3, the same longitudinal peak position is obtained (Figure S1). The two resonance energies were mapped locally resulting in the two images at the center of Figure 3a. The longitudinal and transverse characters can be clearly identified and agree with previous EELS measurements of gold nanorods.^{36,37,39}

The EELS measurements of the deformed particles were taken and plotted in the same way as described above for the undeformed ones. Figure 3b shows the measurement of the deformed particle that is marked as particle 1 in Figure 2. The longitudinal plasmon resonance was found to be blue-shifted to 1.8 eV (690 nm), which corresponds to the observed decrease in aspect ratio with respect to the undeformed particles that exhibited a resonance at 1.55 eV (800 nm). The transverse plasmon resonance stayed at 2.4 eV. The plasmon maps for both energies still show a clear longitudinal and transverse character that is preserved after deformation. It can furthermore be seen that the flatter side of the rod exhibits a less pronounced signal than the tip-shaped end of the rod. This is in agreement with the lightning rod effect and is reproduced by FDTD simulations presented in Figure S3 where the longitudinal resonance (signal at 694 nm) is depicted. As in the experiments, it exhibits a stronger local field enhancement at the round tip of the rod and a less enhanced one at the flat end. Because of the sharp corners in the simulations, the two corners at the flat side also show strong local field enhancements. For the experimental rod, these corners were more rounded and not as sharp and thus no such strong enhancements were observed at these corners.

The EELS measurement results for particle 2 of Figure 2 are shown in Figure 3c. The longitudinal plasmon resonance was shifted to 2.25 eV (551 nm) and starts to overlap with the transverse one. The plasmon maps still exhibit a longitudinal and transverse character but the spatial extent of both resonances cannot be as nicely resolved as for Figure 3a,b due to the limited energy and spatial resolution. As observed for the particles in Figure 3b the flatter end of the rods have a less pronounced plasmonic field enhancement than the

rounded tip of the rods. This difference, however, is less pronounced in Figure 3c due to the overlap of the longitudinal and transverse resonances, which makes it more difficult to separate the two signals for this smaller aspect ratio. As expected, for a complete deformation into a nanosphere only one plasmon resonance at 2.4 eV (520 nm) was observed. As is quite clear from the excitation maps, this resonance is spread out over the whole particle.

Our methodology allows us to not only observe changes in the particles aspect ratio and shape but also to make estimates of the volumes before and after deformation. We estimated the volume from the 2D shape and sizes that we measure from the TEM images. We hereby allow a $\pm 15\%$ discrepancy when comparing the volume before and after laser irradiation due to uncertainties that occur when measuring the diameter and length of the nanorod from TEM images and from small irregularities in shape of the deformed particles. It needs to be mentioned that an exact volume determination is only possible in 3D. 3D tomography has been done on gold nanorods and makes it possible to determine the exact volume.⁶³ For undeformed rods, the known geometry was used to determine the volume. For deformed particles, the volume could be estimated assuming that due to the 3D confinement of the silica shell the particles deform similarly in the third dimension.

For absorbed energies above the melting threshold, particles were found to lose part of their volume during laser irradiation. At 8.1 mJ/cm², only a very small fraction of particles were reduced in size, whereas at 10.6 mJ/cm² about one-third of the particles suffered from a size reduction. Those particles were not included in the analysis because a volume loss can influence the aspect ratio change. Again, it needs to be stressed that this is only a rough estimate because we calculate the volume based on a 2D projection. Examples of particles that underwent more severe size reductions are shown in Figure 4a,

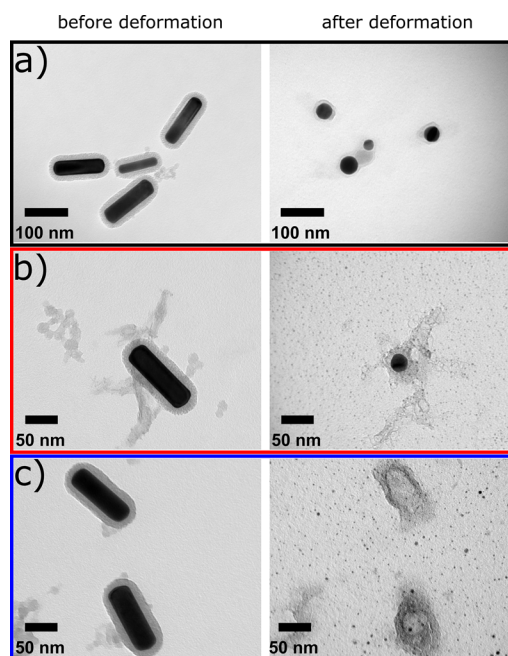


Figure 4. Au volume loss at higher fluences. (a,b) Particles that reduced in size after being illuminated with a fluence of 10.6 mJ/cm². For a fluence of 13.8 mJ/cm², most particles disappeared (c). (b,c) Examples of cluster formation that increases with increasing fluence.

where all volumes dropped to less than one-third of the volumes of the original rods, based on the 2D projection. A size reduction and laser ablation of gold nanoparticles upon laser irradiation is well-known in literature,^{64–66} as well as for fs-pulse irradiation where they used micelle-stabilized rods with a volume 26 times smaller than ours.²⁸ Those experiments were performed at higher fluences (about 200 J/cm²), though, despite the smaller volume. Thus, the size reduction needs to be taken into account for photothermal applications as we found it to also happen for increasing percentages of particles at lower fluences despite the mesoporous silica coating. Such effects would be hard to detect without our single particle analysis where the same particles are observed before and after deformation. At fluences above 10.6 mJ/cm², all particles suffered from size reduction or completely disappeared as shown in Figure 4c. It does not necessarily mean that the particles that disappeared were heated above the boiling temperature of gold (around 3100 K). Fragmentation can happen at temperatures below the melting point even.⁶⁷ At high temperatures the silica shell can also break and release the gold.⁴⁹

Particles that were reduced in size were often accompanied by the presence of small gold droplets on the surface of the TEM grid that can be seen in Figure 2 already and is more extremely visible in Figure 4b,c. This strengthens the point made by calculating the volumes of the particles from the 2D projections that the particles lose volume. It should be stressed that these small particles are most likely not present in the Au NR solution as the TEM images before deformation prove. We verified that the droplets are gold by high-resolution TEM (Figure S4). The volume of clusters found on the grid was found to increase for higher fluences, not necessarily in the number of clusters but rather in their size.

In principle, the mesopores of the silica shell are big enough to allow material transport. Thus, it seems that a lot of gold penetrated the shell through the mesopores and was deposited around the particle or transported away in the surrounding liquid. To check whether that could be the case we coated a small part of the sample with a more dense Stoeber silica shell on top of the mesoporous shell closing off the pores and resulting in a total shell thickness of about 19 nm. Thus, closer to the rod the shell was more mesoporous than at the outer part of the shell. We observed that small gold fragments ended up inside the shell (Figure S5) even before a significant volume reduction took place. Thus, the mesoporosity of the shell does not only allow for accessibility of the surface as is important, for example, for catalysis and drug release, but also for transport of gold fragments out of the shell. However, this happens only at high laser fluences and there is a whole range of parameters for which the fs-illumination gives great control over the single particle aspect ratio without changing the volume of the particle.

In conclusion, we performed single particle deformation experiments on silica-coated gold nanorods under fs-illumination analyzing both shape and plasmon resonances by electron microscopy. Silica-coated rods were found to be more stable compared to uncoated rods and deform via an intermediate bullet-like shape where one end of the rod was found to detach from the mesoporous silica shell and become flatter than the other end that remained attached to the silica shell. These morphological changes were also reflected in the plasmonic properties, which were confirmed by FDTD calculations taking the observed shape as input. A less pronounced longitudinal

resonance and thus local field enhancement was measured and calculated at the flatter side. Furthermore, size reduction of the particles was observed, also for a certain percentage of particles that were not completely deformed yet and at relatively low fluences. Our measurements pave the way toward understanding the deformation behavior of single gold nanorods under a mesoporous silica shell constraint and are therefore of importance for all the applications where a shell enhances properties and specifically for our future project in which we want to tailor the plasmonic resonances of single particles in 3D photonic assemblies in similar ways as a DVD disk can be written in a 2D fashion.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.nanolett.5b04851.

Methods, measured and calculated extinction spectra, estimate of heat dissipation time, confocal point spread functions, FDTD simulations, more details on size reduction, and cluster analysis. (PDF)

■ AUTHOR INFORMATION

Corresponding Authors

*E-mail: W.Albrecht@uu.nl.

*E-mail: A.vanBlaaderen@uu.nl.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We thank Dr. Nicolas Gauquelin for his assistance during the EELS measurements and Thomas Atlantzidis for the high-resolution images of the gold clusters. We furthermore thank Ernest van der Wee for the simulation of the confocal point spread functions. The authors acknowledge financial support from the European Research Council under the European Unions Seventh Framework Programme (FP-2007-2013)/ERC Advanced Grant Agreement 291667 HierarSACol and the Foundation of Fundamental Research on Matter (FOM), which is part of The Netherlands Organisation for Scientific Research (NWO). The authors furthermore acknowledge financial support from the European Research Council (ERC Starting Grant 335078-COLOURATOMS). The authors also appreciate financial support from the European Union under the Seventh Framework Program (Integrated Infrastructure Initiative N. 262348 European Soft Matter Infrastructure, ESMI). This work was supported by the Flemish Fund for Scientific Research (FWO Vlaanderen) through a postdoctoral research grant to B.G.

■ REFERENCES

- (1) Pérez-Juste, J.; Pastoriza-Santos, I.; Liz-Marzán, L. M.; Mulvaney, P. *Coord. Chem. Rev.* **2005**, 249, 1870–1901.
- (2) Huang, X.; Neretina, S.; El-Sayed, M. A. *Adv. Mater.* **2009**, 21, 4880–4910.
- (3) Zijlstra, P.; Orrit, M. *Rep. Prog. Phys.* **2011**, 74, 106401.
- (4) Vigderman, L.; Khanal, B. P.; Zubarev, E. R. *Adv. Mater.* **2012**, 24, 4811–4841.
- (5) Li, N.; Zhao, P.; Astruc, D. *Angew. Chem., Int. Ed.* **2014**, 53, 1756–1789.
- (6) Li, P.-C.; Huang, S.-W.; Wei, C.-W.; Chiou, Y.-C.; Chen, C.-D.; Wang, C.-R. *C. Opt. Lett.* **2005**, 30, 3341–3343.

- (7) Li, P.-C.; Wang, C.-R. C.; Shieh, D.-B.; Wei, C.-W.; Liao, C.-K.; Poe, C.; Jhan, S.; Ding, A.-A.; Wu, Y.-N. *Opt. Express* **2008**, *16*, 18605–18615.
- (8) Son, M.; Lee, J.; Jang, D. J. *J. Mol. Catal. A: Chem.* **2014**, *385*, 38–45.
- (9) Takahashi, H.; Niidome, Y.; Yamada, S. *Chem. Commun.* **2005**, 2247–2249.
- (10) Sudeep, P. K.; Joseph, S. T. S.; Thomas, K. G. *J. Am. Chem. Soc.* **2005**, *127*, 6516–6517.
- (11) Li, C.-Z.; Male, K. B.; Hrapovic, S.; Luong, J. H. T. *Chem. Commun.* **2005**, 3924–3926.
- (12) Alvarez-Puebla, R. A.; Agarwal, A.; Manna, P.; Khanal, B. P.; Aldeanueva-Potel, P.; Carbó-Argibay, E.; Pazos-Pérez, N.; Vigderman, L.; Zubarev, E. R.; Kotov, N. A.; Liz-Marzán, L. M. *Proc. Natl. Acad. Sci. U. S. A.* **2011**, *108*, 8157–8161.
- (13) Sivapalan, S. T.; Devetter, B. M.; Yang, T. K.; van Dijk, T. V.; Schulmerich, M. V.; Carney, P. S.; Bhargava, R.; Murphy, C. J. *ACS Nano* **2013**, *7*, 2099–2105.
- (14) Mansuripur, M.; Zakharian, A. R.; Lesuffleur, A.; Oh, S.-H. H.; Jones, R. J.; Lindquist, N. C.; Im, H.; Kobaykov, A.; Moloney, J. V. *Opt. Express* **2009**, *17*, 14001–14.
- (15) Taylor, A. B.; Kim, J.; Chon, J. W. M. *Opt. Express* **2012**, *20*, 5069–5081.
- (16) Ullah, A.; Li, X.; Cheng, X.; Hao, X.; Su, Y.; Ma, J.; Gu, M. *Opt. Express* **2012**, *20*, 24516.
- (17) Zijlstra, P.; Chon, J. W. M.; Gu, M. *Nature* **2009**, *459*, 410–413.
- (18) Kennedy, L. C.; Bickford, L. R.; Lewinski, N. A.; Coughlin, A. J.; Hu, Y.; Day, E. S.; West, J. L.; Drezek, R. A. *Small* **2011**, *7*, 169–183.
- (19) Akiyama, Y.; Mori, T.; Katayama, Y.; Niidome, T. *Nanoscale Res. Lett.* **2012**, *7*, 565.
- (20) Huang, X. H.; El-Sayed, I. H.; Qian, W.; El-Sayed, M. A. *J. Am. Chem. Soc.* **2006**, *128*, 2115–2120.
- (21) Zhang, Z.; Wang, L.; Wang, J.; Jiang, X.; Li, X.; Hu, Z.; Ji, Y.; Wu, X.; Chen, C. *Adv. Mater.* **2012**, *24*, 1418–1423.
- (22) Murphy, C. J.; Sau, T. K.; Gole, A. M.; Orendorff, C. J.; Gao, J.; Gou, L.; Hunyadi, S. E.; Li, T. *J. Phys. Chem. B* **2005**, *109*, 13857–13870.
- (23) Ni, W.; Kou, X.; Yang, Z.; Wang, J. *ACS Nano* **2008**, *2*, 677–686.
- (24) Juvé, V.; Cardinal, M. F.; Lombardi, A.; Crut, A.; Maioli, P.; Pérez-Juste, J.; Liz-Marzán, L. M.; Del Fatti, N.; Vallée, F. *Nano Lett.* **2013**, *13*, 2234–2240.
- (25) Link, S.; El-Sayed, M. A. *J. Phys. Chem. B* **1999**, *103*, 8410–8426.
- (26) Hu, M.; Wang, X.; Hartland, G. V.; Salgueiriño Maceira, V.; Liz-Marzán, L. M. *Chem. Phys. Lett.* **2003**, *372*, 767–772.
- (27) Chang, S.-S.; Shih, C.-W.; Chen, C.-D.; Lai, W.-C.; Wang, C. R. *C. Langmuir* **1999**, *15*, 701–709.
- (28) Link, S.; Burda, C.; Mohamed, M. B.; Nikoobakht, B.; El-Sayed, M. A. *J. Phys. Chem. A* **1999**, *103*, 1165–1170.
- (29) Link, S.; Burda, C.; Nikoobakht, B.; El-Sayed, M. A. *J. Phys. Chem. B* **2000**, *104*, 6152–6163.
- (30) Mohamed, M. B.; Ismail, K. Z.; Link, S.; El-Sayed, M. A. *J. Phys. Chem. B* **1998**, *102*, 9370–9374.
- (31) Al-Sherbini, A. S. A. M. *Colloids Surf., A* **2004**, *246*, 61–69.
- (32) Petrova, H.; Perez Juste, J.; Pastoriza-Santos, I.; Hartland, G. V.; Liz-Marzán, L. M.; Mulvaney, P. *Phys. Chem. Chem. Phys.* **2006**, *8*, 814–821.
- (33) Link, S.; El-Sayed, M. A. *J. Chem. Phys.* **2001**, *114*, 2362–2368.
- (34) Zijlstra, P.; Chon, J. W. M.; Gu, M. *Phys. Chem. Chem. Phys.* **2009**, *11*, 5915–5921.
- (35) Taylor, A. B.; Siddiquee, A. M.; Chon, J. W. M. *ACS Nano* **2014**, *8*, 12071–12079.
- (36) Bosman, M.; Keast, V. J.; Watanabe, M.; Maarof, A. I.; Cortie, M. B. *Nanotechnology* **2007**, *18*, 165505.
- (37) N'Gom, M.; Li, S.; Schatz, G.; Erni, R.; Agarwal, A.; Kotov, N.; Norris, T. B. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2009**, *80*, 113411.
- (38) Schaffer, B.; Riegler, K.; Kothleitner, G.; Grogger, W.; Hofer, F. *Micron* **2009**, *40*, 269–273.
- (39) Schaffer, B.; Hohenester, U.; Trügler, A.; Hofer, F. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2009**, *79*, 041401.
- (40) Rodríguez-González, B.; Attouchi, F.; Cardinal, M. F.; Myroshnychenko, V.; Stephan, O.; García de Abajo, F. J.; Liz-Marzán, L. M.; Kociak, M. *Langmuir* **2012**, *28*, 9063–9070.
- (41) Myroshnychenko, V.; Nelayah, J.; Adamo, G.; Geuquet, N.; Rodríguez-Fernández, J.; Pastoriza-Santos, I.; MacDonald, K. F.; Henrard, L.; Liz-Marzán, L. M.; Zheludev, N. I.; Kociak, M.; García De Abajo, F. J. *Nano Lett.* **2012**, *12*, 4172–4180.
- (42) Nicoletti, O.; de la Peña, F.; Leary, R. K.; Holland, D. J.; Ducati, C.; Midgley, P. A. *Nature* **2013**, *502*, 80–84.
- (43) Tollan, C. M.; Marcilla, R.; Pomposo, J. A.; Rodríguez, J.; Aizpurua, J.; Molina, J.; Mecerreyes, D. *ACS Appl. Mater. Interfaces* **2009**, *1*, 348–352.
- (44) Liu, Y.; Mills, E. N.; Composto, R. J. *J. Mater. Chem.* **2009**, *19*, 2704–2709.
- (45) Khalavka, Y.; Ohm, C.; Sun, L.; Banhart, F.; Sonnichsen, C. *J. Phys. Chem. C* **2007**, *111*, 12886–12889.
- (46) Li, J. F.; Huang, Y. F.; Ding, Y.; Yang, Z. L.; Li, S. B.; Zhou, X. S.; Fan, F. R.; Zhang, W.; Zhou, Z. Y.; Wu, D. Y.; Ren, B.; Wang, Z. L.; Tian, Z. Q. *Nature* **2010**, *464*, 392–395.
- (47) Sekhar, A. C. S.; Meera, C. J.; Ziyad, K. V.; Gopinath, C. S.; Vinod, C. P. *Catal. Sci. Technol.* **2013**, *3*, 1190–1193.
- (48) Chen, J.; Zhang, R.; Han, L.; Tu, B.; Zhao, D. *Nano Res.* **2013**, *6*, 871–879.
- (49) Gergely-Fülöp, E.; Zámbo, D.; Deák, A. *Mater. Chem. Phys.* **2014**, *148*, 909–913.
- (50) Chen, Y.-S.; Frey, W.; Kim, S.; Homan, K.; Kruizinga, P.; Sokolov, K.; Emelianov, S. *Opt. Express* **2010**, *18*, 8867–8878.
- (51) Chen, L.-C.; Wei, C.-W.; Souris, J. S.; Cheng, S.-H.; Chen, C.-T.; Yang, C.-S.; Li, P.-C.; Lo, L.-W. *J. Biomed. Opt.* **2010**, *15*, 016010.
- (52) Zijlstra, P.; Chon, J. W. M.; Gu, M. *Opt. Express* **2007**, *15*, 12151–12160.
- (53) Chen, Y. S.; Frey, W.; Kim, S.; Kruizinga, P.; Homan, K.; Emelianov, S. *Nano Lett.* **2011**, *11*, 348–354.
- (54) Besseling, T. H.; Hermes, M.; Kuijk, A.; de Nijs, B.; Deng, T. S.; Dijkstra, M.; Imhof, A.; van Blaaderen, A. *J. Phys.: Condens. Matter* **2015**, *27*, 194109.
- (55) Myroshnychenko, V.; Rodríguez-Fernández, J.; Pastoriza-Santos, I.; Funston, A. M.; Novo, C.; Mulvaney, P.; Liz-Marzán, L. M.; García de Abajo, F. J. *Chem. Soc. Rev.* **2008**, *37*, 1792–1805.
- (56) Baffou, G.; Quidant, R. *Laser Photonics Rev.* **2013**, *7*, 171–187.
- (57) Prescott, S. W.; Mulvaney, P. *J. Appl. Phys.* **2006**, *99*, 123504.
- (58) Plech, A.; Kotaidis, V.; Grésillon, S.; Dahmen, C.; Von Plessen, G. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2004**, *70*, 195423.
- (59) Link, S.; Burda, C.; Nikoobakht, B.; et al. *Chem. Phys. Lett.* **1999**, *315*, 12–18.
- (60) van Kats, C. M. *Anisotropic Model Colloids*. Ph.D. Thesis, Utrecht University, Utrecht, Netherlands, 2008.
- (61) Besseling, T. H.; Jose, J.; van Blaaderen, A. *J. Microsc.* **2015**, *257*, 142–150.
- (62) Chon, J. W. M.; Bullen, C.; Zijlstra, P.; Gu, M. *Adv. Funct. Mater.* **2007**, *17*, 875–880.
- (63) Goris, B.; Bals, S.; Van den Broek, W.; Carbó-Argibay, E.; Gómez-Graña, S.; Liz-Marzán, L. M.; Van Tendeloo, G. *Nat. Mater.* **2012**, *11*, 930–935.
- (64) Mafuné, F.; Kohno, J. Y.; Takeda, Y.; Kondow, T. *J. Phys. Chem. B* **2001**, *105*, 9050–9056.
- (65) Inasawa, S.; Sugiyama, M.; Koda, S. *Jpn. J. Appl. Phys.* **2003**, *42*, 6705–6712.
- (66) Muto, H.; Miyajima, K.; Mafuné, F. *J. Phys. Chem. C* **2008**, *112*, 5810–5815.
- (67) Plech, A.; Kotaidis, V.; Lorenc, M.; Boneberg, J. *J. Nat. Phys.* **2006**, *2*, 44–47.